## <sup>81</sup>Br and <sup>127</sup>I NQR of Rare Earth Trihalogenides REX<sub>3</sub>, X = Br, I\*

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The  $^{81}Br$  NQR spectra of REBr<sub>3</sub>, RE = Dy, Ho, Yb, Sm, and the  $^{127}I$  NQR spectra of REI<sub>3</sub>, RE = Dy, Ho, Sm, are reported. Additionally  $^{127}I$  NQR data of RbHoI<sub>4</sub>, KSmI<sub>4</sub>, and Rb<sub>3</sub>Sm<sub>2</sub>I<sub>9</sub> are given. The line shape of the  $^{127}I$  NQR of HoI<sub>3</sub> was studied in external magnetic fields up to 300 Gauss.

## Introduction

Trihalogenides of rare earth elements show some differences in their crystal lattices. The chlorides and bromides of the light rare earth elements LaCl<sub>3</sub>...GdCl<sub>3</sub>; LaBr<sub>3</sub>...PrBr<sub>3</sub>, are isomorphous to UCl<sub>3</sub> [1]. The trichlorides of the heavy rare earth elements are isomorphous to AlCl<sub>3</sub> and the corresponding tribromides crystallize with the FeCl<sub>3</sub> type structure [2, 3]. Several tribromides REBr<sub>3</sub> (RE = Nd, Eu, Sm, Tb) and the iodides REI<sub>3</sub> (RE = La, Ce, Pr, Nd) are isomorphous to PuBr<sub>3</sub>, whereas the majority of the REI<sub>3</sub> shows the BiI<sub>3</sub> type structure [4, 5].

NQR is quite a sensitive method with respect to details in the structure of the electrons surrounding the resonating nuclei, and changes in the crystal structure can easily be detected.

## **Results and Discussion**

We have studied several tribromides and triiodides of rare earth elements, and a few compounds formed by rare earth triiodides and alkali iodides. In Table 1 the  $^{81}$ Br and  $^{127}$ I NQR frequencies are listed for T = 77 K (in a few cases for 290 K, too).

Since the compounds DyBr<sub>3</sub>, HoBr<sub>3</sub>, and YbBr<sub>3</sub>, are isomorphous to FeCl<sub>3</sub> ( $C_{3i}^2 - R\bar{3}$ , Z = 6), a single line <sup>81</sup>Br NQR spectrum is expected and the experimental results are in accordance with this expecta-

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tion (see Table 1).  $v(^{81}\text{Br})$  increases with decreasing distance M-Br (d(Dy-Br) = 2.836 Å; d(Ho-Br) = 2.825 Å); d(Yb-Br) = 2.798 Å). A small electric field gradient, EFG, is expected at the site of the rare earth ion since RE is surrounded by 6 Br-atoms in almost regular octahedral coordination.

SmBr<sub>3</sub> crystallizes in the orthorhombic system (D<sub>2h</sub><sup>17</sup>-Ccmm, Z = 4) [3] and the Br-atoms occupy the point positions 4 c and 8 f.

The <sup>81</sup>Br NQR spectrum is in agreement with the conclusion which follows from the crystal structure. Two <sup>81</sup>Br lines are found, the higher frequency line corresponding to the Br atoms at 4 c; the lower frequency line (position 8 f) shows double intensity.

Sometimes ago Parks and Moulton [6] studied NdBr<sub>3</sub> by NQR. They observed two resonance frequencies for <sup>81</sup>Br and determined the asymmetry parameter of both EFG tensors. At the side 4 c they found  $\eta = 6.8\%$  and at the site 8 f  $\eta = 48.7\%$ .

The rare earth triiodides studied by us have the BiI<sub>3</sub> type structure. As can be seen from Table 1, the asymmetry parameter of the EFG at the iodine site

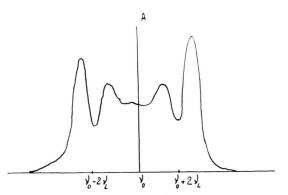


Fig. 1. The line form of NQR  $^{127}$ I ( $\Delta m = 3/2 - 5/2$ ) in the polycrystalline sample of HoI<sub>3</sub> located in the external magnetic field  $3 \cdot 10^{-2}$  T.  $v_{\rm L} = {\rm Larmor\ frequency}$ .

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Table 1. NOR frequencies of <sup>81</sup>Br and <sup>127</sup>I, quadrupole coupling constants,  $e^2 \Phi_{-1} O h^{-1}$  (<sup>127</sup>I), and asymmetry parameters  $\eta(^{127}I)$ , in some trihalogenides of the rare earth elements.  $\nu(^{81}Br)$  at

Substance	$v(^{81}\mathrm{Br})/\mathrm{MHz}$				
DyBr <sub>3</sub> HoBr <sub>3</sub> YbBr <sub>3</sub> SmBr <sub>3</sub>	30.36 31.194 34.224 24.985 46.665				
Substance	$v(^{127}I)/MHz,$ $\Delta m = 1/2 - 3/2$	$v(^{127}I)/MHz$ , $\Delta m = 3/2 - 5/2$	$\frac{e^2  \Phi_{zz}  Q  h^{-1}}{\text{MHz}}$	η (%)	T/K
DyI <sub>3</sub>	42.15 42.24	83.33 82.80	276.7 278.3	9.5 13.5	77 290
HoI <sub>3</sub>	43.46 43.50	85.84 85.23	286.7 286.8	10.0 14.0	77 290
$SmI_3$	36.72	72.76	242.5	9.0	300
RbI·HoI <sub>3</sub>	38.19	74.30	248.2	15.0	77
$KI \cdot SmI_3$	41.09	80.64	270.3	12.5	77
$3RbI \cdot 2SmI_3$	34.01 46.49	63.29 86.94	214.4 294.1	24.0 23.0	77

is rather small ( $\approx 10\%$ ). This points out that there is a small deviation of the iodine positions from the ideal BiI<sub>3</sub> type structure in which the point symmetry of the iodine would be 3 ( $\eta \equiv 0$ ).

For SmI<sub>3</sub> the <sup>127</sup>I NQR frequency  $(1/2 \rightleftharpoons 3/2)$ was found to increase very little with decreasing temperature ( $\sim 0.012 \, \text{kHz/degree}$ ). In the range  $110 \le T/K \le 120$  the intensity of the line becomes very weak and at T = 77 K no <sup>127</sup>I NQR signal was detected in the range  $50 \le v/MHz \le 150$ .

 $HoI_3$  shows a very strong transition  $3/2 \rightleftharpoons 5/2$  and the signal to noise ratio is 100 on the oscilloscope. The <sup>127</sup>I Zeeman spectrum on polycrystalline material was studied at 77 K in a field  $B_0 = 3 \cdot 10^{-2} \,\mathrm{T}$  $(B_0 || B_{HF})$ . It was found that the shape of the band is very similar to that found for Sb in Sb<sub>2</sub>O<sub>3</sub> and Re in NaReO<sub>4</sub> [7], but different from the shape expected in case of negligible dipole-dipole interactions. This effect can be observed by comparing the <sup>35</sup>Cl Zeeman NQR powder spectrum of NaClO<sub>3</sub> and KClO<sub>3</sub>, where in the latter compound dipolar interactions are very small.

The 127I NQR Zeeman powder spectrum  $(3/2 \rightleftharpoons 5/2)$  of SmI<sub>3</sub> is shown in Figure 1. It seems to be of interest to elaborate the theory of NQR line shape by incorporating dipole-dipole interactions.

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